

The MSDS format adheres to the standards and regulatory requirements of the United States and may not meet regulatory requirements in other countries. DuPont 1 Page Material Safety Data Sheet COLOR MATCH SPRAYS 6108FR Revised 22-MAY-1998 CHEMICAL PRODUCT/COMPANY IDENTIFICATION Company Identification MANUFACTURER/DISTRIBUTOR DuPont Chemical Solutions Enterprise 1007 Market Street Wilmington, DE 19898 PHONE NUMBERS Product Information : 1-800-441-7515 Transport Emergency : CHEMTREC 1-800-424-9300 Medical Emergency : 1-800-441-3637 Medical Emergency \_\_\_\_\_ COMPOSITION/INFORMATION ON INGREDIENTS \_\_\_\_\_ Components % CAS Number Material Acetone 67-64-1 20-25 Aliphatic Hydrocarbon 64742-89-8 30-35 64742-48-9 Mineral Spirits 0-5 \*Xylene 1330-20-7 0-5 68476-86-8 Propane-Isobutane Mixture 30 \* Disclosure as a toxic chemical is required under Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR part 372. \_\_\_\_\_ HAZARDS IDENTIFICATION Potential Health Effects Skin contact can cause drying of the skin, irritation with discomfort or rash, or dermatitis. Xylene can be absorbed through the skin in amounts capable of causing toxic effects. Skin contact to the propane/isobutane mixture may cause frostbite. Eye contact can cause severe irritation, redness, tearing and blurred vision. Intentional misuse by deliberately concentrating and inhaling the contents may be harmful or fatal. Inhalation may cause irritation of the nose and throat with sneezing, sore throat or runny nose or asphyxiation (the amount of oxygen available for breathing is reduced)

## (HAZARDS IDENTIFICATION - Continued)

Inhalation of the propane/isobutane mixture may cause irregular heartbeat with a strange sensation in the chest, "heart thumping", apprehension, lightheadedness, feeling of fainting, dizziness, weakness, sometimes progeressing to loss of consciousness and death.

Gross inhalation overexposure to xylene may cause cardiovascular effects, impaired functioning of the blood-forming system with alterations in blood cell counts and/or anemia, pathological changes in the liver and kidneys and fatality. Repeated exposures by inhalation has occassionally caused skin sensitization in humans.

Ingestion or inhalation of the mixture may cause central nervous system depression with dizziness, confusion, incoordination, drowsiness or unconsciousness; or non-specific effects such as headache, nausea and weakness.

Ingestion of the mixture can cause gastrointestinal irritation, nausea, vomiting and diarrhea. A serious ingestion hazard is aspiration (liquid entering the lungs during ingestion or vomiting) which may result in "chemical pneumonia". Symptoms include coughing, gasping, choking, shortness of breath, bluish discoloration of the skin, rapid breathing and heart rate, and fever. Pulmonary edema or bleeding, drowsiness, confusion, coma and seizures may occur in more serious cases. Symptoms may develop immediately or as late as 24 hours after the exposure, depending on how much chemical entered the lungs.

Reports have associated repeated and prolonged occupational overexposure to solvents with permanent brain and nervous system damage; and adverse effects to kidneys, lungs, and spleen, heart and adrenals. Exposures to certain aliphatic hydorcarbons have been associated with polyneuropathy.

Increased susceptibility to the effects of Xylene (Mixed Isomers) may be observed in persons with pre-existing disease of the central nervous system, kidneys, liver, cardiovascular system, lungs, or bone marrow.

#### Carcinogenicity Information

None of the components present in this material at concentrations equal to or greater than 0.1% are listed by IARC, NTP, OSHA or ACGIH as a carcinogen.

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# FIRST AID MEASURES First Aid

#### INHALATION

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.

SKIN CONTACT

Flush area with lukewarm water. Do not use hot water. If frostbite has occurred, call a physician.

EYE CONTACT

In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Call a physician.

#### INGESTION

If swallowed, do not induce vomiting. Immediately give 2 glasses of water. Never give anything by mouth to an unconscious person. Call a physician.

Notes to Physicians

THIS MATERIAL MAY MAKE THE HEART MORE SUSCEPTIBLE TO ARRHYTHMIAS. Catecholamines such as adrenaline, and other compounds having similar effects, should be reserved for emergencies and then used only with special caution.

Because of the danger of aspiration, emesis or gastric lavage should not be employed unless the risk is justified by the presence of additional toxic substances. Activated charcoal may be given but should be used with caution since it may induce vomiting.

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FIRE FIGHTING MEASURES
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Flammable Properties

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Flash Point: 0 F (-18 C)Method: Tag Closed Cup - TCC.Flammable limits in Air, % by VolumeLEL: 1 %
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Extremely flammable.

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## (FIRE FIGHTING MEASURES - Continued)

Closed containers may explode and/or autoignite when exposed to extreme heat, vapors are heavier than air and may travel along the ground or may be moved by ventilation and ignited by pilot lights, other flames, sparks, heaters, smoking, electric motors or other locations distant from material handling point.

Extinguishing Media

Alcohol Resistant Foam, Dry Chemical, CO2. Water may be ineffective but should be used to cool fire exposed structures and containers.

If water is used, fog muzzles are preferable. Use self-contained breathing apparatus with full face piece operated in pressure demand or other positive pressure mode.

ACCIDENTAL RELEASE MEASURES

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Safeguards (Personnel)

NOTE: Review FIRE FIGHTING MEASURES and HANDLING (PERSONNEL) sections before proceeding with clean-up. Use appropriate PERSONAL PROTECTIVE EQUIPMENT during clean-up.

Spill Clean Up

Eliminate all ignition sources, ventilate area, absorb liquid on paper, vermiculite, floor absorbent, or other absorbent material and transfer to a closed container.

Material collected on absorbent material may be deposited in a posted toxic substance landfill in accordance with local, State, and Federal regulations.

HANDLING AND STORAGE

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Handling (Personnel)
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Do not inhale. Avoid contact with eyes, skin or clothing.

Handling (Physical Aspects)

Do NOT puncture. Keep away from heat, sparks and flames.

Do NOT incinerate.

#### Storage

Store away from heat, sparks and flames, sunlight.

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(HANDLING AND STORAGE - Continued)

Do NOT store in areas above 120 deg F. Store large quantities in building protected for storage of flammable liquids.

EXPOSURE CONTROLS/PERSONAL PROTECTION

Personal Protective Equipment

EYE/FACE PROTECTION

Wear coverall chemical splash goggles when possibility exists for eye and face contact due to splashing or spraying material.

RESPIRATORS

Wear NIOSH approved respiratory protection, as appropriate.

PROTECTIVE CLOTHING

Wear impervious clothing, such as gloves, apron, boots, or whole bodysuit as appropriate.

# Exposure Guidelines

Applicable Exposure Limits Acetone PEL. (OSHA) : 1000 ppm, 2400 mg/m3, 8 Hr. TWA : 500 ppm, 1188 mg/m3, A4 (ACGIH) TLV STEL 750 ppm, 1782 mg/m3, A4 AEL \* (DuPont) : 500 ppm, 8 & 12 Hr. TWA Mineral Spirits : None Established  $\mathbf{PEL}$ (OSHA) TLV (ACGIH) : None Established AEL \* (DuPont) : 100 ppm, 8 Hr. TWA Xylene  $\mathbf{PEL}$ (OSHA) : 100 ppm, 435 mg/m3, 8 Hr. TWA TLV (ACGIH) : 100 ppm, 434 mg/m3, 8 Hr. TWA, A4 STEL 150 ppm, 651 mg/m3, A4 : 100 ppm, 8 & 12 Hr. TWA 150 ppm, 15 minute TWA AEL \* (DuPont)

\* AEL is DuPont's Acceptable Exposure Limit. Where governmentally imposed occupational exposure limits which are lower than the AEL are in effect, such limits shall take precedence.

\_\_\_\_\_ PHYSICAL AND CHEMICAL PROPERTIES \_\_\_\_\_ Physical Data Boiling Point : 133-331 F (56-166 C) Melting Point : 1500 F (816 C) Estimated minimum Vapor Pressure : Unknown for products Solubility in Water : Slight to moderate Vapor Density : Heavier than air Specific Gravity : <1 (Water=1.0) Evaporation Rate : Slower than ether % Volatiles : 80-90 % Valatile Operation Contents (55 % Volatile Organic Content: 65 % Appearance and Odor: Typical solvent paintWeight per gallon: 6.5-7.5 (PAINT) \_\_\_\_\_ STABILITY AND REACTIVITY \_\_\_\_\_ Chemical Stability Conditions to Avoid Heat, sparks, and flames. Incompatibility with Other Materials Incompatible or can react with strong oxidizing agents and Decomposition May form toxic materials, carbon dioxide/carbon monoxide, various hydrocarbons, nitrogen compounds, etc., when burned. Polymerization Polymerization will not occur. TOXICOLOGICAL INFORMATION \_\_\_\_\_ Animal Data

ACETONE

Stable.

heat.

Oral LD50, rat: 9,750 mg/kg Inhalation 4 hour, LC50, rat: 31,983 ppm Dermal LD50, rabbit: 20,000 mg/kg

Animal testing indicates that Acetone is an eye irritant, but is not a skin irritant. Acetone has not been tested for (TOXICOLOGICAL INFORMATION - Continued)

skin sensitization in animals.

Repeated dermal exposure of animals caused dry skin and cataracts. Long-term dermal exposure caused no significant toxicological effects.

Repeated ingestion exposure to high doses of acetone caused kidney injury, reduced weight gain, and liver, hematological and testicular effects.

Single and repeated exposure by inhalation to high doses caused central nervous system depression, and decreased motor activity. Repeated exposures to higher concentrations caused incoordination and reduced weight gain.

In animal testing Acetone has not caused carcinogenicity. Limited data on the exposure of pregnant rats to Acetone show developmental toxicity only at exposure levels producing other toxic effects in the adult animal. Limited data on the exposure of pregnant mice to Acetone show a reduction of fetal body weight and an increase in the incidence of late resorptions.

The NOEL (No-Observed-Effect-Level) for developmental toxicity in the rat and mouse study was 2200 ppm. Limited data on the exposure of rats and mice to Acetone show reproductive toxicity only at exposure levels producing other toxic effects in the adult animal. Acetone does not cause genetic damage in bacterial cells. Test in mammalian cell cultures have been both positive and negative. Testing in yeast has also produced positive results.

MINERAL SPIRITS

Inhalation 4 hour LC50: > 1,400 ppm in rats Skin absorption LD50: > 5 mL/kg in rabbits Oral LD50: > 5 mL/kg in rats

The compound is a mild skin irritant, is an eye irritant, and is untested for animal sensitization.

Ingestion: A single large dose caused lethargy, diarrhea and incoordination.

Inhalation: A single exposure caused eye, nose and throat irritation with slight incoordination. Repeated exposure produces respiratory irritation and kidney damage (in male rats only)

Adequate tests for carcinogenic activity have not been performed. Tests in animals demonstrate no developmental or reproductive toxicity.

The material does not produce genetic damage in bacterial or

## DuPont Material Safety Data Sheet

## (TOXICOLOGICAL INFORMATION - Continued)

mammalian cell cultures or in tests in animals. It does not produce heritable genetic damage.

XYLENE

Oral LD50:4,500 mg/kg in ratsDermal ALD:4,320 mg/kg (>5 mL/kg) in rabbitsInhalation 4 hour LC50:6,700 ppm in rats

Animal testing indicates Xylene is an eye irritant and a moderate to severe skin irritant. Single dermal exposure to high doses of Xylene caused narcosis.

Single exposure to Xylene by ingestion caused prostration and incoordination. Repeated exposure caused shallow respiration, prostration, liver and kidney effects, reduced weight gain, and altered hematology and clinical chemistry. Long-term exposure caused decreased body weight and histopathological changes of the liver.

Single exposure to Xylene by inhalation caused upper respiratory tract irritation, behavioral effects, incoordination, prostration, altered respiratory rate, low blood pressure, and altered hematology. Repeated exposure caused incoordination; impaired sense of hearing; histopathological changes of the liver, kidneys, adrenals, heart, spleen, lungs, and bone marrow; altered hematology; and reduced weight gain. Long-term exposure caused liver effects.

One published study reports limited data suggesting high oral doses caused an increase in malignant tumors in rats. However, other more extensive animal studies have demonstrated no evidence of carcinogenicity. Animal data show developmental effects only at or near levels producing other toxic effects in the adult animal. There is a report in the literature that indicates synergistic developmental effects with Xylene and Acetylsalicylic Acid. Reproductive data on adult animals show no change in reproductive performance. Tests have shown that Xylene does not cause genetic damage in bacterial or mammalian cell cultures, or in animals. In animal testing, Xylene has not caused permanent genetic damage in reproductive cells of mammals (has not produced heritable genetic damage). 8

\_\_\_\_\_ ECOLOGICAL INFORMATION \_\_\_\_\_ Ecotoxicological Information MINERAL SPIRITS The compound is slightly toxic. The 48 hr LC50 in Golden Orfe: 320-435 mg/L. \_\_\_\_\_ TRANSPORTATION INFORMATION Shipping Information DOT Proper Shipping Name : Consumer Commodity Hazard Class : ORM-D I.D. No. (UN/NA) : None Packing Group : None DOT Label(s) : None DOT/IMO Proper Shipping Name : Paint Hazard Class : 3.2 : 1263 UN No. Packing Group : II DOT/IMO Label : Flammable Liquid IATA/ICAO Proper Shipping Name : Consumer Commodity Hazard Class : 9 UN No. : ID8000 Packing Group : None : Miscellaneous Label OTHER INFORMATION \_\_\_\_\_ NFPA, NPCA-HMIS NFPA Rating : 2 Health Flammability : 4 : 0 Reactivity NPCA-HMIS Rating : 1 Health : 4 Flammability : 0 Reactivity

Personal Protection rating to be supplied by user depending on use conditions.

## (Continued)

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The data in this Material Safety Data Sheet relates only to the specific material designated herein and does not relate to use in combination with any other material or in any process.

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# Indicates updated section.

This information is based upon technical information believed to be reliable. It is subject to revision as additional knowledge and experience is gained.

End of MSDS